

### CHARACTERIZATION AND PREPARATION OF (AL<sub>2</sub>O<sub>3</sub>) NANO FLUIDS FOR SOLAR THERMAL APPLICATIONS

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#### Abstract

This work is attempted to prepare nanofluids for solar thermal Application like Solar Still, Solar Water Heater etc. For that Alumina (Al2O3) Nanofluids with different weight concentration level have been tested for zeta potential, particle size of dispersion by using Horiba Zeta Sizer and Absorbency bv using UVvis-Spectrophotometer. Thermal conductivity of the prepared Nanofluids have been calculated using KD2 Thermal property Analyzer and compared the thermal conductivity of the various concentration of nanofluids.

Keywords: Condensation, Nanofluids, Stabilty, thermal conductivity.

### I. INTRODUCTION

Nanofluids are a new class of fluids engineered by dispersing nanometer-sized materials (nanoparticles, nanofibers, nanotubes, nanowires, nanorods, nanosheet, or droplets) in base fluids. In other words, nanofluids are nanoscale colloidal suspensions containing condensed nanomaterials. They are two-phase systems with one phase (solid phase) in another (liquid phase). Nanofluids have been found to possess enhanced thermophysical properties such as thermal conductivity, thermal diffusivity, convective viscosity, and heat transfer coefficients compared to those of base fluids like oil or water. It has demonstrated great potential applications in many fields. For a two-phase system, there are some important issues we have to face. One of the most important issues is the stability of nanofluids, and it remains a big challenge to achieve desired stability of nanofluids. In this paper, we will review the new

progress in the methods for preparing stable summarize the nanofluids and stability mechanisms. In recent years, nanofluids have attracted more and more attention. The main driving force for nanofluids research lies in a wide range of applications. Although some review articles involving the progress of nanofluid investigation were published in the past several years [1–6], most of the reviews are concerned of the experimental and theoretical studies of the thermophysical properties or the convective heat transfer of nanofluids. The purpose of this paper will focus on the new preparation methods and stability mechanisms, especially the new application trends for nanofluids in addition to the heat transfer properties of nanofluids. We will try to find some challenging issues that need to be solved for future research based on the review on these aspects of nanofluids.

### II. CHARACTERIZATION OF NANOFLUIDS

### A. PREPARATION OF NANOFLUIDS

Preparation of nanofluids is the key step in the use of nanoparticles to improve the thermal conductivity of fluids. Generally, nanoparticles are hydrophobic in nature, prone to agglomerate together and settled quickly. To maintain a stable and even suspension surfactants such as sodium dodecyl benzene sulphonate (SDBS) is used. fig.1 [8]. At first 0.1g of SDBS is dissolved in 50 ml water in two beakers separately in that calculated amount of Al2o3 nanoparticles (fig.2) is added for 0.05% and 0.1% concentrations. The prepared samples were stirred by using Magnetic Stirrer shown in figure.3 for 15 Minutes and ultrasonicated for 1 hours using Ultrasonicator shown in figure. 4. Then the pH value of solution is maintained as

8 by adding HCL and NaOH to get maximum thermal conductivity. [9]. The pH value of Nanofluids after preparation is tabulated in table.1. The amount of Nanoparticles to be added is calculated by the formula

Amount of Nano Particle in GMS = concentration (%)/100 x water sample taken x density of the Nano Particle.



FIG.1 SDBS



FIG.2 (AL203)

Nano materi al	Density (g/cm <sup>3</sup> )	Amountofnanoparticlesaddedwith50mlofwater(gms)		pH Adjustme nt
		0.05%	0.1%	
$Al_2O_3$	3.95	0.0987	0.197	8
		5	5	

### Table 1 Nanofluids Sample Preparationdetails.

Nanomateria	рН		
1	0.05 %	0.1%	
Aluminium Oxide(Al <sub>2</sub> O <sub>3</sub> )	5	6	

Table.2 pH Value of Nanofluids Sample afterpreparation

The pH value of the of the prepared Nanofluids sample were measured. The pH

value of the Aluminum oxide for 0.1% of concentration is 5 and for 0.1% of concentration are 6. It is adjusted to 8 to get good thermal conductivity.

# **B. STABILTY EVALUATION METHODS**

The prepared Nanofluids are then evaluated for stability to choose the good concentration which gives better stability.

Concentration (%)	Zeta potential(mV)	Particle size (nm)	Best concentration	
0.05	-38.7	58	- 0.1%	
0.1	-52.3	50		

Table.3. Stability results

## C. ZETA POTENTIAL AND PARTICLE SIZE ANALYSIS

The zeta potential and particle size of the dispersion in the nanofluids is measured by using Horiba zeta Analyzer. The results of the Nano fluids are shown table 3. The figure 3 and Figure 4 shows zeta potential results for 0.1% and 0.05% concentration Respectively.

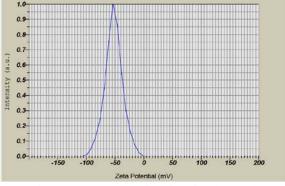


Fig 3 Zeta Potential Al<sub>2</sub>O<sub>3</sub> (0.1%)

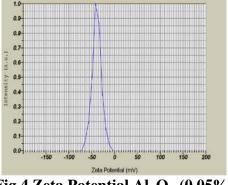
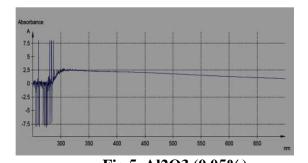


Fig 4 Zeta Potential Al<sub>2</sub>O<sub>3</sub> (0.05%)

### **D. UV-VIS-SPECTROPHOTOMETER:**

Then the prepared Nanofluids have been tested for Absorbency by using UV-vis-spectrophotometer. The fig.5 and fig.6 shows result obtained from uvvis-spectrophotometer for 0.05% and 0.1 % concentration levels. The result shows 0.1% of Al2O3 has good Absorbency level compared to 0.05% of concentration.



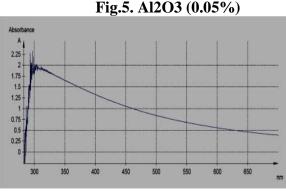


Fig.6. Al2O3 (0.1%)

### E. PHOTOGRAPHIC METHOD

In final stage the stability is evaluated bySolar thermal Applications. Hence in this work photographic method. The figure 5 and 6 show Characterization of Nano fluids have been done. photographs of nanofluid samples after 2 hours and n future we can directly use of Nano fluids in seven days of preparation respectively. After sevensolar thermal Applications without any days there is some visible sedimentation in the characterization work. bottom of the carrying bottle.



**Fig.7.After two hours** Al2o3 (0.05%) Al2o3 (0.1%)



Fig.8.After 7days III THERMAL CONDUCTIVITY RESULTS

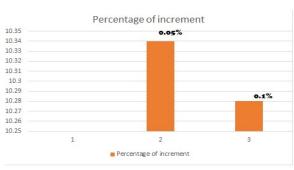
After tested Zeta Potential and Particle size it shows good concentration of Nanofluids to be tested with the Nanofluids. The 0.1% concentration of Nanofluids is tested for Thermal conductivity by using KD2 Thermal property Analyzer. The Results are tabulated in table 4.6

Nanofluids	Concentration	Thermal Conductivity W/m.K	Percentage of increment
$Al_2O_3$	0.1%	0.6355	10.34
$Al_2O_3$	0.05%	0.6255	10.28

Table.4 thermal conductivity results

### **IV RESULTS**

The Thermal conductivity of the Al2o3 have been calculated and it is concluded that the 0.1% concentration is very much suitable for



**Fig.8.Percentage of Increment** 

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