

SYNTHESIS, CHARACTERIZATION AND BIOLOGICAL EVALUATION OF 1-HEPTA-O-BENZOYL-B-D-MALTOSYL-3-SUBSTITUTED BENZOTHIAZOLYL CARBAMIDES

Anuja M. Mopari*, Shirish P.Deshmukh, Gajanan V. Korpe P.G.Department of chemistry, Shri Shivaji collage,Akola-444001,(M.S.) India E-mail:anujamopari@gmail.com

Abstract:

1-hepta-O-benzoyl-β-D-**Synthesis** of maltosyl-3-substituted benzothiazolyl carbamides by the interaction of 1-hepta-Obenzoyl-β-D-maltosyl Isocvanate with various substituted benzothiazoles. The compoundshave synthesized been characterized by spectral technique such as FTIR,NMR and Mass and usual chemical transformations. The newly synthesized compounds were screened for their microbial activities.

Keywords: Isocyanate, Maltosylated, Benzothiazolyl, Carbamides, Antimicrobial Activity.

1)Introduction:-

Benzothiazoles are bicyclic ring system having multiple applications. Substituted benzothiazoles have found applications in several areas of chemistry.2aminobenzothiazoles are broadly found in bioinorganic medicinal chemistry with applications in drug discovery and development for the treatment of diabetes¹,epilepsy²,inflammation³,amyotropic lateral sclerosis⁴, analgesia⁵, tuberculosis⁶ and viral infections⁷.It is also a novel mechanistic class of antitumor agents⁸⁻⁹.Moreover, they were studied extensively for their activities like antiglutamate, immunosupressive¹⁰, amyloid agents¹¹ imaging and thrombin inhibitors¹².Furthermore, the importance of substituted benzothiazoles is well established in the protection of paper against forgery¹³ and agricultural chemistry. Apart from this several methodologies have been reported for the preparation of benzothiazoles¹⁴. As isocyanates are the most powerful starting materials for the synthesis of various N-linked sugar derivatives.

Taking in to account previous knowledge in this field, we have synthesized series of maltosyl benzothiazolyl carbamides by the addition of maltosvl Isocyanate the substituted to benzothiazoles.The reaction of molecular bromine with aryl thiocarnamides is known to 2-aminobenzothiazoles.herein produce we report he Synthesis of new 1-hepta-O-benzoylβ-D-maltosyl-3-substituted benzothiazolyl carbamides (III_{a-f}) by the interaction of 1-hepta-*O*-benzoyl-β-D-maltosyl Isocvanate with various substituted benzothiazoles.

2)Materials and method 2.1 Materials

The melting point of compounds were determined with the help of Thermionic melting point apparatus and were found uncorrected. The structures of newly synthesized compounds were confirmed on the basis of elemental and spectral analysis. IR Spectra were recorded on KBr disks on SHIMADZU IR affinity-1 FTIR spectrometer. ¹H NMR was obtained on Bruker DRX-300 NMR Spectrometer. Samples were prepared in CDCl₃ with TMS as an internal reference. The mass spectra were obtained on JEOL-AccuTof JMS-T100LC and Thermo Fennigan LCQ Advantage max ion trap mass spectrometer. Thin layer chromatography (TLC) was performed on silica gel G for TLC (Merck) and spot were visualized by iodine vapour.

2.2 Method: The reagents required for the given synthesis are obtained as follows-

2.2.1:Preparation ofhepta-*O*-Benzoyl-β-D-Maltosyl Isocyanates (I):

Hepta-O-Benzoyl- β -D-Maltosyl Isocyanateswas prepared by the condensation of Hepta-O-Benzoyl- α -D-Maltosyl bromide (0.005 mol 3.4g) and lead cyanate (0.005 mol, 1.4g) in

INTERNATIONAL JOURNAL OF CURRENT ENGINEERING AND SCIENTIFIC RESEARCH (IJCESR)

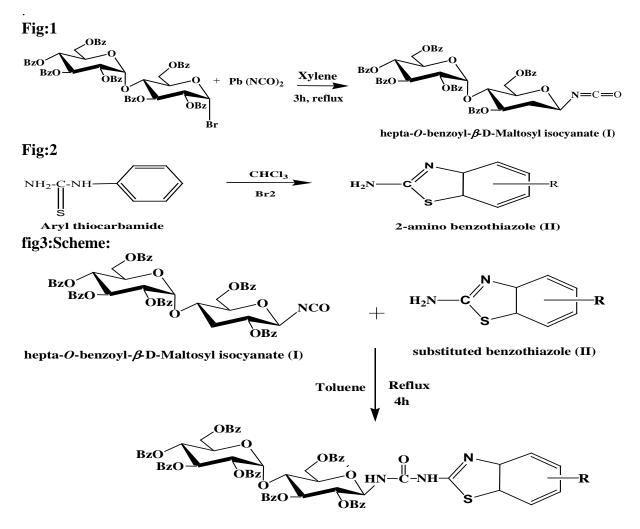
boiling xylene (25 ml) medium for3 hrs. with frequent shaking. After the removal of lead bromide the xylene filtrate was triturated with petroleum ether (60-80°C) when Hepta-*O*-Benzoyl- β -D-Maltosyl Isocyanates was precipitated out. It was purified by dissolving it in a minimum quantity of chloroform and reprecipitating with petroleum ether (60-80°C) to afford a pale yellow solid. The homogeneity of the product was checked by TLC. m.p. 112 °C,(Fig:1)

2.2.2: Preparation of substituted benzothiazolyl carbamides (II):

The reaction of molecular bromine with aryl thiocarbamides is known to produce 2aminobenzothiazole so the requisitesubstituted benzothiazoles were prepared by oxidative cyclization of substituted aryl thiocarbamides with molecular bromine in chloroform.(Fig:2)

Synthesisof1-hepta-O-benzoyl- β -D-maltosyl-3-substituted benzothiazolyl carbamides(III_{a-f})

When a mixture of hepta-*O*-benzoyl-β-Dmaltosyl isocyanate(I) and 2-aminobezothiazole (II) was refluxed in toluene medium for 4 hr. After completion of the reaction, the solvent get distilled off and sticky residue obtained was triturated with petroleum ether gives a light yellow colour solid. It was crystallized by ethanol-water. The purity of compound was checked by TLC. The spectral analysis IR, NMR, Mass spectra of the product were observed. Optical rotation of the product were also recorded in table-1. All the compounds have been screen for microbial activity were recorded in table-2



hepta-O-benzoyl- β -D-Maltosyl 3-Substituted benzothiazolyl carbamides.(IIIa-f) Where, Bz-COC₆H₅,R'= a) H, b)CH₃,c) chloro

2.3:Spectral and Elemental analysis¹⁵⁻¹⁷:-1-heptaO-benzoyl- β -D-Maltosyl -3-(2)benzothiazolyl carbamide. (IIIa) **IR** (**KBr cm-1**): 3034 (ArC-H), 2850 (Ali C-H), 1732(C=O),1651 (C=N), 1271 (C-N), 1070 (Characte-ristic maltose), 709 (C-S).

INTERNATIONAL JOURNAL OF CURRENT ENGINEERING AND SCIENTIFIC RESEARCH (IJCESR)

1H NMR (CDCl3) δ: 8.09-7.26 (39 H,

m,Aromatic protons), 7.61 (1H, s, NH proton), 6.72 (1.3H, s, NH proton), 6.20 (14H, m, maltosyl protons)

Mass *m/z*: 1245(M+), 1217, 1175, 1007, 975, 133, 105.

1-hepta -*O*-benzoyl- β -D-Maltosyl -3-(2)-4methyl benzothiazolyl carbamides. (IIIb)

IR(KBr cm-1): 2956 (Ar C-H), 2858 (Ali C-H), 1732 (C=O),1651 (C=N), 1256 (C-N), 1070 (Characteristic maltose), 709 (C-S).)

1H NMR (CDCl3) δ: 8.06-7.23 (38 H, m, Aromatic protons), 6.72 (1H, s, NH proton), 5.73 (1H, s, NH proton), 2.49(3H, s, methyl protons), 4.56-4.45 (14H, m, maltosyl protons) **Mass** *m/z***:** 1259(M+), 1245, 1217, 769, 651, 163, 191.

1-hepta-*O*-benzoyl β -D-Maltosyl -3-(2)-4chloro benzothiazolyl carbamides (IIId)

IR (**KBr cm-1**):3034 (Ar C-H), 2898 (Ali C-H), 1728 (C=O),1647(C=N), 1271 (C-N), 1070 (Characteristic maltose), 709 (C-S).

1H NMR (CDCl3) δ: 8.05-7.26 (38 H, m, Aromatic protons), 6.72 (1H, s, NH proton), 6.17 (1H, s, NH proton), 7.3-4.28 (14H, m,

maltosyl protons)

proton),7.3-4.28 (14H, m, maltosyl protons) **Mass** *m*/*z*: 1279(M+), 1247, 1217, 1177, 581, 185.

2.4: Microbial Assay:

The microbial assay of synthesized compounds have been studied using cup plate agar diffusion method¹⁸⁻¹⁹ by measuring the inhibition zone in mm. the compounds were taken at a concentration of 1 mg/ml using dimethyl sulphoxide (DMSO) as solvent.

The bacterial assay of compounds wasstudied against Escherichia coli, Р. vulgaris Staphylococcus aureus and Pseudomonas aeruginosa in nutrient agar medium. Amikacin µg/ml) was used as standard for (100)antibacterial activity. The results are presented in Table 2. The compounds were screen for antifungal activity against Aspergillusnigar and Candida albicance in potato dextrose agar medium *flucanzole* (100µg/ml) as standard for antifungal activity.Result were recorded in Table 2.

3.Result and discussion:

All products were crystallized from ethanolwater before recording the physical data (table-1).The purity of compounds was checked by TLC The spectral analysis IR,¹HNMR,Masspectra of the product were observed. Optical rotation of the product was also recorded.Yeild and M.P.of compound were obtained satisfactory.Antimicrobial activity of newly synthesised compound were discussed in table-2 as below.

Table1: Characterization of 1-hepta-O-benzoyl- β -D-maltosyl-3-substituted benzothiazolyl carbamides.

Sr. No.	Synthesized Compounds	M.P. (⁰ C)	Yield (%)		Analysis (%) Required)	Rf Values	$[\alpha]^{28}{}_{\rm D}$ [c, in CHCl ₃]
				N	S		
1.	IIIa	160 ⁰ C	71 %	3.80(3.37)	2.48(2.57)	0.61	+105 ⁰
2.	IIIb	180 ⁰ C	67%	3.30(3.33)	2.47(2.54)	0.58	+60.32 ⁰
3.	IIIc	185 ^o C	60 %	3.31(3.33)	2.49(2.54)	0.57	$+40.0^{\circ}$
4.	IIId	210 ⁰ C	67 %	3.10 (3.28)	2.32(2.50)	0.66	+104.06 ⁰
5.	IIIe	190 ⁰ C	63%	3.00(3.28)	2.1(2.50)	0.62	+64.56 ⁰
6.	IIIf	195 ⁰ C	69%	3.18(3.28)	2.31(2.50)	0.63	+85.33 ⁰

C and H analysis were found satisfactory in all cases.

Sr.	Compounds		Anti	Antifungal ^{**}			
No.		E. coli	P.vulgaris	S. aureus	Ps. Aeruginosa	C. albicance	A. Niger
1.	IIIa	21	13	22	12	23	15
2.	IIIb	15	12	24	18	25	13
3.	IIIc	20	12	25	16	21	13
4.	IIId	13	21	10	13	22	14
5.	IIIe	15	10	14	15	20	16
6.	IIIf	17	15	13	14	20	19
7.	Amikacin	21	23	28	24	-	-
8.	Fluconazole	-	-	-	-	26	24

carbamides

**zone of inhibition in mm (15 or less) resistance, (16-20mm) moderate and (more than 20mm) sensitive. Escherichia coli (E. coli), (P.Vulgaris), Proteus vulgaris Staphalococcusaureus(S. aureus), Psudomonasauriginosa (*Ps*. auriginosa), Candida albicance (*C.albicance*) and Aspergillusniger (A.niger).

Antibacterial study of these compound indicated that IIIa and IIIc were found to be active against *E. coli* and rest of were found to be moderately active.CompoundIIId exhibited more significant activity against*P.Vulgaris and* IIIa,IIIb,IIIc against *S. aureus*.CompoundIIIb,IIIc shows moderate against *Ps.Aeruginosa*. All other compounds exhibited low to moderate activity.

The results of **antifungal** activities are also tabulated in Table-2.Almost all compounds shows promising activity against *C.albicance* and compound IIIe and IIIf exhibited moderate activity against *A.niger and all* other compounds are resistant.

Conclusion:

of new benzothiazolyl А series carbamides were successfully synthesized and characterized by IR, ¹H NMR and Mass and Elemental analysis. Spectral The synthesized compounds were evaluated for their antibacterial activities. Thus. the newly benzothiazolyl synthesized carbamides derivatives, exhibits comparable antibacterial activities against the organisms tested. The method adopted in this investigation is simple, efficient and inexpensive and is useful in

synthesizing pharmacologically important molecules.

Acknowledgement:-

Authors are thankful to SAIF/CIL, Panjab University, Chandigarh, Central Instrumental Cell, Shri Shivaji College, Akola for providing spectral data and alsothankful to Dr.R. M. Bhise, Principal, Shri Shivaji College, Akola for providing necessary facilities.

References:

- 1) H.Suter,H.Zutter (1967)*Helv.Chim.Acta.*,50,1084.
- S.J.Hays,M.J.Rice,D.F.Ortwine,G.Johns on,R.D.Schwartz.K.Boyd,L.F.Copeland, M.G.Vartanian,P.A.Boxer (1994)*J.Pharma.Sci.*,83,1425.
- 3) S.N.Sawhney,S.K.Arora,J.V.Singh,O.P. Banal,S.P.Singh(1978)*IndianJ.Chem.*,16 B 605.
- 4) G.Bensimon,L.Lacomblez,V.Meininger(1994) *New Engl.J.Med.*,330,585.
- 5) G.Foscolos,G.Tsatsas,A.Champagnac,M .Pommier(1977)*Ann.Pharma.Fr*, 35, 295.
- 6) Mauro Vieira de Almeida, Mireille Le Hyaric, Giovanni Wison Amarante, Maria Cristina Silva Lourenco, Marcelo Luiz Lima Brandao (2007) *Eur. J. Med. Chem.*, 42, 1076-1083.
- 7) C.J. Paget,K. Kisner, R. L. Stone, D. C. Delong (1969)..*J.Med.Chem.*, 12.1016.
- 8) I.aleta,M.Kralji,M.Marjanovi,B.Bertoa,S .Tomi,G.Pavlovi,K.Paveli,Grace Karminski-Zamola(2009) *J.Med.Chem.*, 52(6),1744-1756.

INTERNATIONAL JOURNAL OF CURRENT ENGINEERING AND SCIENTIFIC RESEARCH (IJCESR)

- 9) S.Aiello,G.Wells,E.L.Stone,H.Kadri,R.B azzi,D.R.Bell,M.F.G.Stevens,C.S.Mattle ws,T.D.Bradshaw,A.D.Westwell(2008) *J.Med.Chem.*,51(16),5135-5139.
- 10) T.Mase, H. Arima, K. Tomioka, T. Yamada, K. Murase: (1986)*J Med.Chem.*,29 (3),386-394.
- C. A. Mathis, Y. Wang, D.P. Holt, G.-F.Huang, M. L. Debnath, W.E.Klunk (2003) J.Med. Chem., 46(13),2740-2754.
- P. Morinko, A.Krbavcic, G.Mlinsek, T. Solmajer, A.T. Bakija, M Stegner, J. Stojan, D. Kikelj(2004)*Eur. J. Med. Chem.*, 39, 257-265.
- 13) W. Kirk Jr., Johnson, A. T. Blomquist (1943) *J.Org. Chem.*, 08(6), 557-563.
- 14) G.W.Stewart, C.A.Baxter, EdCleator, Fay eJ.Sheen (2009) J. Org. Chem., 74(8), 3229-3231.

- 15) Silverstein RM, Bassler GC, and Morrill TC, 1991, "Spectrometric Identification of organic compounds", 5thed, Wiley, New York.
- 16) Buclzikiewicz H, Djerassi C and Williams DH, 1964, "Structural Elucidation of natural product by mass spectrometry", *II Holden Day*, 207.
- 17) Jacobsen NE, 2007, NMR Spectroscopy Explained, Wiley, 15.
- 18) Kawangh F, 1963, Analytical Microbiology, Academic press, New York.
- 19) British pharmacopoeia- II, Biological assay and Tests, The Stationary Office Ltd., London, A-205, 1998.